

3-(2-Bromophenyl)isocoumarin

Aamer Saeed,^a Edwin F. van der Eide^b and Masood Parvez^{b*}^aDepartment of Chemistry, Quaid-i-Azam University, Islamabad 45320, Pakistan, and^bDepartment of Chemistry, University of Calgary, 2500 University Drive NW, Calgary, Alberta, Canada T2N 1N4

Correspondence e-mail: parvez@ucalgary.ca

Key indicators

Single-crystal X-ray study

T = 173 K

Mean $\sigma(\text{C}-\text{C}) = 0.006 \text{ \AA}$

R factor = 0.038

wR factor = 0.094

Data-to-parameter ratio = 15.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

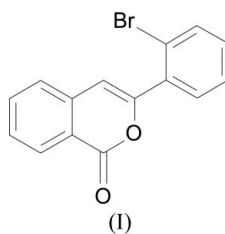
The molecular structure of the title compound, $\text{C}_{15}\text{H}_9\text{BrO}_2$, consists of two essentially planar units, benzopyran-1-one and 2-bromophenyl, which are inclined at $51.42(12)^\circ$ with respect to one another. The structure is stabilized by two weak intermolecular $\text{C}-\text{H}\cdots\text{O}$ interactions.

Received 21 June 2006

Accepted 28 June 2006

Comment

More than two hundred isocoumarins and 3,4-dihydroisocoumarins have been isolated, predominantly from a variety of fungi, lichens and bacteria, and to a lesser extent from higher plants, insects and marine organisms, and the number of known isocoumarins is still increasing dramatically (Barry, 1964; Napolitano, 1997). Isocoumarins are useful intermediates (Hauser & Baghdanor, 1988; Mali & Babu, 1998) in the synthesis of a variety of natural products. We have reported the syntheses of a number of naturally occurring and synthetic isocoumarins (Saeed, 2003*a,b*, 2004*a,b*; Saeed & Ehsan, 2005). 3-Halophenylisocoumarins are not known in nature and are expected to display a number of bioactivities. The title compound, (I), was prepared in order to investigate its bioactivity systematically. In this paper, the structure of (I) is described.



The structure of (I) consists of an essentially planar benzopyran-1-one unit, the maximum deviation of any atom

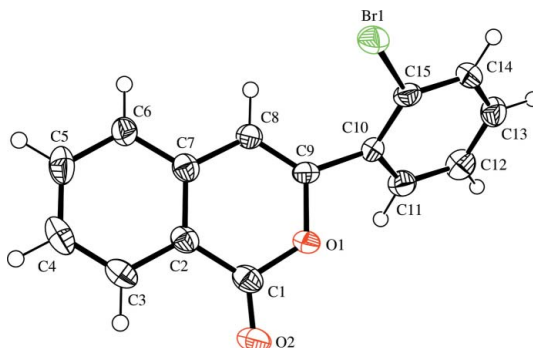


Figure 1
ORTEP (Johnson, 1976) drawing of (I), with displacement ellipsoids plotted at the 50% probability level. H atoms are drawn as small spheres.

from its plane being 0.025 (4) Å for C1, and a 2-bromophenyl unit for which Br1 is 0.046 (8) Å out of the plane of the benzene ring (Fig. 1); the mean planes of the two units are inclined at 51.42 (12)° with respect to one another. The molecular dimensions in (I) agree with the corresponding dimensions reported for 3,4-dihydroisocoumarins and isocoumarins included in the Cambridge Structural Database (Version 5.27, 2005 Release; Allen, 2002). The structure is stabilized by two weak intermolecular C—H...O interactions (Fig. 2 and Table 1). There are no significant π — π stacking interactions.

Experimental

A stirred mixture of homophthalic acid (0.5 g, 2.77 mmol) and 2-bromobenzoyl chloride (2.12 g, 9.7 mmol) was heated in an oil bath at 473 K for 4 h. Thin-layer chromatography of the residue (petroleum ether–ethyl acetate 8:3) followed by recrystallization from MeOH–H₂O (4:1) gave the title isocoumarin (0.7 g, 2.35 mmol, 85%) as light-yellow needles (m.p. 391–393 K). ¹H NMR (CDCl₃, δ , p.p.m.): 6.90 (s, 1H, H-4), 7.43 (m, 1H, H-5), 7.47 (m, 1H, H-7), 7.49 (m, 1H, H-4'), 7.51 (m, 1H, H-6'), 7.69 (dt, J = 7.2, 1.8 Hz, 1H, H-6); 7.81 (dt, J = 7.8, 1.7 Hz, 1H, H-5').

Crystal data

C ₁₅ H ₉ BrO ₂	$Z = 4$
$M_r = 301.13$	$D_x = 1.701 \text{ Mg m}^{-3}$
Orthorhombic, <i>Pna</i> ₂₁	Mo $K\alpha$ radiation
$a = 12.595$ (6) Å	$\mu = 3.48 \text{ mm}^{-1}$
$b = 12.128$ (9) Å	$T = 173$ (2) K
$c = 7.699$ (6) Å	Block, colourless
$V = 1176.0$ (14) Å ³	$0.16 \times 0.14 \times 0.10 \text{ mm}$

Data collection

Bruker–Nonius KappaCCD diffractometer	7105 measured reflections
ω and φ scans	2492 independent reflections
Absorption correction: multi-scan (SORTAV; Blessing, 1997)	2096 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.594$, $T_{\max} = 0.708$	$R_{\text{int}} = 0.056$
	$\theta_{\text{max}} = 27.5^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0492P)^2 + 0.803P]$
$R[F^2 > 2\sigma(F^2)] = 0.038$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.094$	$(\Delta\sigma)_{\text{max}} = 0.001$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.37 \text{ e } \text{Å}^{-3}$
2492 reflections	$\Delta\rho_{\text{min}} = -0.50 \text{ e } \text{Å}^{-3}$
164 parameters	Absolute structure: Flack (1983), 1073 Friedels
H-atom parameters constrained	Flack parameter: 0.536 (17)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C4—H4...O2 ⁱ	0.95	2.55	3.439 (6)	155
C6—H6...O2 ⁱⁱ	0.95	2.45	3.242 (7)	140

Symmetry codes: (i) $x - \frac{1}{2}, -y + \frac{1}{2}, z$; (ii) $-x + \frac{1}{2}, y + \frac{1}{2}, z - \frac{1}{2}$.

The crystal is inversion twinned with roughly equal components. The Friedel pairs (1073) were not merged during the refinement. H atoms were located in difference Fourier syntheses and were included

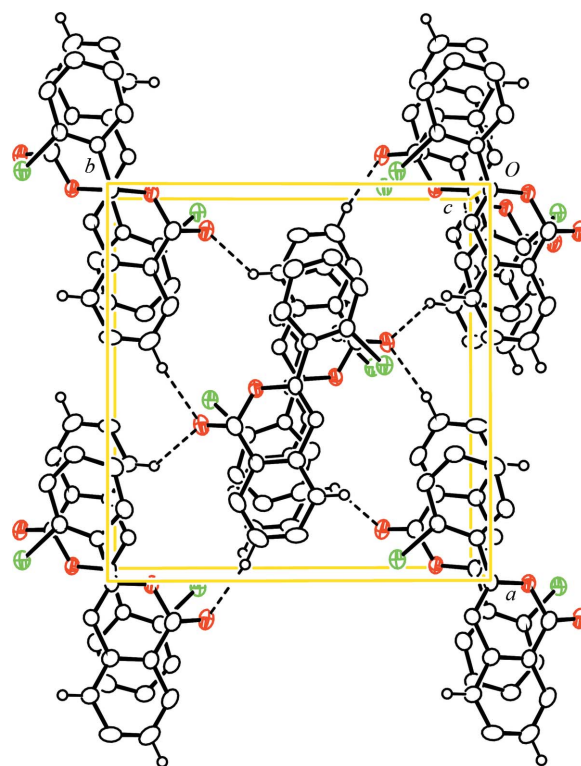


Figure 2

ORTEP (Johnson, 1976) drawing of the packing of (I), showing weak intermolecular C—H...O hydrogen bonds as dashed lines. H atoms not involved in hydrogen bonding have been omitted.

in the refinement at geometrically idealized positions, with C—H = 0.95 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: COLLECT (Hooft, 1998); cell refinement: DENZO (Otwinowski & Minor, 1997); data reduction: SCALEPACK (Otwinowski & Minor, 1997); program(s) used to solve structure: SAPI91 (Fan, 1991); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP (Johnson, 1976); software used to prepare material for publication: SHELXL97 (Sheldrick, 1997).

References

- Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.
 Barry, R. D. (1964). *Chem. Rev.* **64**, 229–260.
 Blessing, R. H. (1997). *J. Appl. Cryst.* **30**, 421–426.
 Fan, H.-F. (1991). *SAPI91*. Rigaku Corporation, Tokyo, Japan.
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
 Hooft, R. (1998). *COLLECT*. Nonius BV, Delft, The Netherlands.
 Hauser, F. M. & Baghdanov, V. M. (1988). *J. Org. Chem.* **53**, 4676–4681.
 Johnson, C. K. (1976). *ORTEP*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
 Mali, R. S. & Babu, K. N. (1998). *J. Org. Chem.* **63**, 2488–2492.
 Napolitano, E. (1997). *Org. Prep. Proced. Int.* **29**, 631–664.
 Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
 Saeed, A. (2003a). *Helv. Chim. Acta*, **86**, 377–383.
 Saeed, A. (2003b). *J. Heterocycl. Chem.* **40**, 337–340.
 Saeed, A. (2004a). *Nat. Prod. Res.* **18**, 373–378.
 Saeed, A. (2004b). *J. Heterocycl. Chem.* **41**, 975–978.
 Saeed, A. & Ehsan, S. (2005). *Chem. Heterocycl. Compd.* **11**, 1644–1648.
 Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.